

Physics Unit 12: The Dual Nature of Light Review

1. Know about double-slit experiment, diffraction, diffraction grating, dispersing light into a spectrum, single-slit diffraction, limits of resolution, blackbody radiation and the relationship between temperature and frequency of light emitted, photoelectric effect, quantization, evidence that light is a wave, evidence that light is a particle, particle-wave duality of nature
2. Why is it difficult to observe everyday sized objects' wave nature?
3. At what angle is the first-order maximum for 800.0-nm wavelength light falling on double slits separated by 0.00100 mm?
4. Calculate the wavelength of light that has its third minimum at an angle of 10.0° when falling on double slits separated by $8.000\ \mu\text{m}$.
5. Light with a 700nm wavelength is shown through a double slit. If the $m = 0$ and $m = 1$ bright fringes are separated by 10° , what is the separation of the slits?
6. A diffraction grating has 2000 lines/cm and has monochromatic light shown on it. If the 3rd-order maximum is at 20° , what is the wavelength of the light?
7. What is the distance between lines on a diffraction grating that produces a second-order maximum for 200.0-nm light at an angle of 20.0° ?
8. Light with a wavelength of 250 nm uniformly illuminates a single slit. What is the width of the slit if the first-order dark fringe is located at $\theta = 1.50^\circ$?
9. Light with a 700nm wavelength is shown through a single slit onto a screen 3 m away. What is the width of the slit if the 2nd-order dark fringe is located 50 cm from the center of the central bright region?
10. Calculate the minimum angular spreading of a laser beam that is originally 1.00 mm in diameter with an average wavelength of 680.0 nm.
11. A spy satellite is in orbit at a distance of 5.0×10^6 m above the ground. It carries a telescope that can resolve the two rails of a railroad track that are 1.0 m apart using light of wavelength 400 nm. What is the diameter of the lens in the telescope?
12. A radio antenna emits photons at a frequency of 101.5 MHz. What is the energy of this photon in Joules?
13. A photon strikes a detector with 2.00 eV of energy. What is the wavelength of the photon?
14. What is the maximum kinetic energy in eV of electrons ejected from a metal by 800-nm EM radiation, given that the binding energy is 0.70 eV?
15. Find the longest-wavelength photon that can eject an electron from a metal, given that the binding energy is 2.00 eV.
16. Find the momentum of a photon with a wavelength of 1200 nm.

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Answers

2. The wavelength is too small to observe.

$$3. \sin \theta = m \frac{\lambda}{d} \rightarrow \sin \theta = 1 \left(\frac{800.0 \times 10^{-9} \text{ m}}{0.00100 \times 10^{-3} \text{ m}} \right) \rightarrow \theta = \mathbf{0.800^\circ}$$

$$4. \sin \theta = m \frac{\lambda}{d} \rightarrow \sin 10.0^\circ = 3 \left(\frac{\lambda}{8.000 \times 10^{-6} \text{ m}} \right) \rightarrow \sin 10.0^\circ = 375000 \lambda \rightarrow \lambda = 4.63 \times 10^{-7} \text{ m} = \mathbf{463 \text{ nm}}$$

$$5. \sin \theta = m \frac{\lambda}{d} \rightarrow \sin 10^\circ = \frac{1(700 \times 10^{-9} \text{ m})}{d} \rightarrow d = 4.03 \mu\text{m} = \mathbf{4.03 \times 10^{-6} \text{ m}}$$

$$6. \sin \theta = m \frac{\lambda}{d}$$

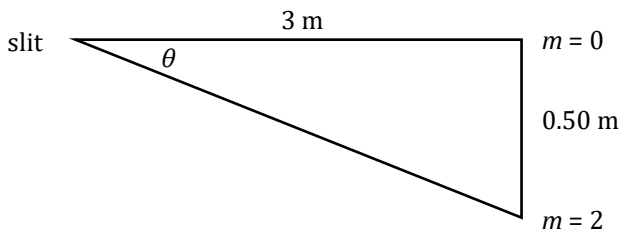
$$d = \frac{1}{2000 \frac{\text{lines}}{\text{cm}}} = 0.0005 \text{ cm} = 0.000005 \text{ m}$$

$$\sin 20^\circ = 3 \left(\frac{\lambda}{0.000005 \text{ m}} \right) \rightarrow \lambda = 5.7 \times 10^{-7} \text{ m} = \mathbf{570 \text{ nm}}$$

$$7. \sin \theta = m \frac{\lambda}{d} \rightarrow \sin 20.0^\circ = 2 \left(\frac{200.0 \times 10^{-9} \text{ m}}{d} \right) \rightarrow d \sin 20.0^\circ = 4.00 \times 10^{-7} \text{ m} \rightarrow d = \mathbf{1.17 \times 10^{-6} \text{ m}}$$

$$8. \sin \theta = m \frac{\lambda}{W} \rightarrow \sin 1.50^\circ = 1 \left(\frac{250 \times 10^{-9} \text{ m}}{W} \right) \rightarrow W \sin 1.50^\circ = 2.50 \times 10^{-7} \text{ m} \rightarrow W = \mathbf{9.55 \times 10^{-6} \text{ m}}$$

$$9. \sin \theta = m \frac{\lambda}{W}$$



$$\tan \theta = \frac{0.5}{3} \rightarrow \theta = 9.46^\circ$$

$$\sin 9.46^\circ = \frac{2(700 \times 10^{-9} \text{ m})}{W} \rightarrow W = \mathbf{8.52 \times 10^{-6} \text{ m}}$$

$$10. \theta = 1.22 \frac{\lambda}{D} \rightarrow \theta = 1.22 \left(\frac{680.0 \times 10^{-9} \text{ m}}{1.00 \times 10^{-3} \text{ m}} \right) \rightarrow \theta = \mathbf{8.30 \times 10^{-4} \text{ rad}}$$

11. Use a right triangle to find the angle in radians:

$$\tan \theta = \frac{1.0 \text{ m}}{5.0 \times 10^6 \text{ m}} \rightarrow \theta = 2 \times 10^{-7} \text{ rad}$$

$$\theta = 1.22 \frac{\lambda}{D} \rightarrow 2 \times 10^{-7} \text{ rad} = 1.22 \left(\frac{400 \times 10^{-9} \text{ m}}{D} \right) \rightarrow$$

$$D(2 \times 10^{-7} \text{ rad}) = 4.88 \times 10^{-7} \text{ m} \rightarrow D = \mathbf{2.44 \text{ m}}$$

$$12. E = nhf \rightarrow E = (1)(6.626 \times 10^{-34} \text{ Js})(101.5 \times 10^6 \text{ Hz}) \rightarrow E = \mathbf{6.72 \times 10^{-26} \text{ J}}$$

$$13. 2.00 \text{ eV} \left(\frac{1.60 \times 10^{-19} \text{ J}}{1 \text{ eV}} \right) = 3.20 \times 10^{-19} \text{ J}$$

$$E = nhf \rightarrow 3.20 \times 10^{-19} \text{ J} = (1)(6.626 \times 10^{-34} \text{ Js})f \rightarrow f = 4.83 \times 10^{14} \text{ Hz}$$

$$c = f\lambda \rightarrow 3.00 \times 10^8 \frac{\text{m}}{\text{s}} = (4.83 \times 10^{14} \text{ Hz})\lambda \rightarrow \lambda = 6.21 \times 10^{-7} \text{ m} = \mathbf{621 \text{ nm}}$$

$$14. KE = \frac{hc}{\lambda} - BE \rightarrow KE = \frac{(6.626 \times 10^{-34} \text{ Js})(3.00 \times 10^8 \frac{\text{m}}{\text{s}})}{800 \times 10^{-9} \text{ m}} \left(\frac{1 \text{ eV}}{1.60 \times 10^{-19} \text{ J}} \right) - 0.70 \text{ eV} \rightarrow KE = 1.55 \text{ eV} - 0.70 \text{ eV} \rightarrow KE = \mathbf{0.85 \text{ eV}}$$

$$15. KE = \frac{hc}{\lambda} - BE \rightarrow 0 = \frac{(6.626 \times 10^{-34} \text{ Js})(3.00 \times 10^8 \frac{\text{m}}{\text{s}})}{\lambda} - 2.00 \text{ eV} \left(\frac{1.60 \times 10^{-19} \text{ J}}{1 \text{ eV}} \right) \rightarrow 3.20 \times 10^{-19} \text{ J} = \frac{1.99 \times 10^{-25} \text{ Jm}}{\lambda}$$

$$\lambda(3.20 \times 10^{-19} \text{ J}) = 1.99 \times 10^{-25} \text{ Jm} \rightarrow \lambda = 6.21 \times 10^{-7} \text{ m} = \mathbf{621 \text{ nm}}$$

$$16. p = \frac{h}{\lambda} \rightarrow p = \frac{6.626 \times 10^{-34} \text{ Js}}{1200 \times 10^{-9} \text{ m}} \rightarrow p = \mathbf{5.52 \times 10^{-28} \text{ kg m/s}}$$

